

Problem 1-5

Before Planck's theoretical work on blackbody radiation, Wien showed empirically that (Equation 1.4)

$$\lambda_{\max} T = 2.90 \times 10^{-3} \text{ m} \cdot \text{K}$$

where $\lambda_{\max}$ is the wavelength at which the blackbody spectrum has its maximum value at a temperature T . This expression is called the Wien displacement law; derive it from Planck's theoretical expression for the blackbody distribution by differentiating Equation 1.3 with respect to λ . *Hint:* Set $hc/\lambda_{\max} k_B T = x$ and derive the intermediate result $e^{-x} + (x/5) = 1$. This problem cannot be solved for x analytically but must be solved numerically. Solve it by iteration on a hand calculator, and show that $x = 4.965$ is the solution.

Solution

The Planck distribution law for blackbody radiation is given by

$$\rho_{\nu}(T) d\nu = \frac{8\pi h}{c^3} \frac{\nu^3}{e^{h\nu/k_B T} - 1} d\nu.$$

Use the chain rule to write this in terms of the wavelength λ .

$$\rho_{\nu}(T) \left(\frac{d\nu}{d\lambda} \right) d\lambda = \frac{8\pi h}{c^3} \frac{\nu^3}{e^{h\nu/k_B T} - 1} \left(\frac{d\nu}{d\lambda} \right) d\lambda \quad (1)$$

The relationship between frequency and wavelength for blackbody radiation is

$$\lambda\nu = c \quad \Rightarrow \quad \begin{cases} \nu = \frac{c}{\lambda} \\ \frac{d\nu}{d\lambda} = -\frac{c}{\lambda^2} \end{cases}.$$

As a result, equation (1) becomes

$$\begin{aligned} [-\rho_{\lambda}(T)] d\lambda &= \frac{8\pi h}{c^3} \frac{\left(\frac{c}{\lambda}\right)^3}{\exp\left(\frac{hc}{k_B T \lambda}\right) - 1} \left(-\frac{c}{\lambda^2}\right) d\lambda \\ &= -\frac{8\pi hc}{\lambda^5} \frac{d\lambda}{\exp\left(\frac{hc}{k_B T \lambda}\right) - 1}. \end{aligned}$$

Therefore, the Planck distribution in terms of wavelength is

$$\rho_{\lambda}(T) d\lambda = \frac{8\pi hc}{\lambda^5} \frac{d\lambda}{\exp\left(\frac{hc}{k_B T \lambda}\right) - 1}.$$

In order to find the wavelength at which the spectrum has its maximum value, differentiate $\rho_{\lambda}(T)$ with respect to λ and set it equal to zero.

$$\frac{d}{d\lambda} \rho_{\lambda}(T) = 0$$

Substitute the distribution and evaluate the derivative.

$$\begin{aligned}
 0 &= \frac{d}{d\lambda} \left[\frac{8\pi hc}{\lambda^5} \frac{1}{\exp\left(\frac{hc}{k_B T \lambda}\right) - 1} \right] \\
 &= 8\pi hc \frac{d}{d\lambda} \left[\frac{1}{\lambda^5} \frac{1}{\exp\left(\frac{hc}{k_B T \lambda}\right) - 1} \right] \\
 &= 8\pi hc \left\{ -\frac{5}{\lambda^6} \frac{1}{\exp\left(\frac{hc}{k_B T \lambda}\right) - 1} + \frac{1}{\lambda^5} \frac{-\exp\left(\frac{hc}{k_B T \lambda}\right) \left(-\frac{hc}{k_B T \lambda^2}\right)}{\left[\exp\left(\frac{hc}{k_B T \lambda}\right) - 1\right]^2} \right\} \\
 &= 8\pi hc \left\{ -\frac{5}{\lambda^6} \frac{\exp\left(\frac{hc}{k_B T \lambda}\right) - 1}{\left[\exp\left(\frac{hc}{k_B T \lambda}\right) - 1\right]^2} + \frac{\lambda}{\lambda^6} \frac{-\exp\left(\frac{hc}{k_B T \lambda}\right) \left(-\frac{hc}{k_B T \lambda^2}\right)}{\left[\exp\left(\frac{hc}{k_B T \lambda}\right) - 1\right]^2} \right\} \\
 &= 8\pi hc \left\{ \frac{-5 \exp\left(\frac{hc}{k_B T \lambda}\right) + 5}{\lambda^6 \left[\exp\left(\frac{hc}{k_B T \lambda}\right) - 1\right]^2} + \frac{\frac{hc}{k_B T \lambda} \exp\left(\frac{hc}{k_B T \lambda}\right)}{\lambda^6 \left[\exp\left(\frac{hc}{k_B T \lambda}\right) - 1\right]^2} \right\} \\
 &= 8\pi hc \frac{\left(-5 + \frac{hc}{k_B T \lambda}\right) \exp\left(\frac{hc}{k_B T \lambda}\right) + 5}{\lambda^6 \left[\exp\left(\frac{hc}{k_B T \lambda}\right) - 1\right]^2}
 \end{aligned}$$

Divide both sides by $8\pi hc$.

$$0 = \frac{\left(-5 + \frac{hc}{k_B T \lambda}\right) \exp\left(\frac{hc}{k_B T \lambda}\right) + 5}{\lambda^6 \left[\exp\left(\frac{hc}{k_B T \lambda}\right) - 1\right]^2}$$

Multiply both sides by $\lambda^6 \left[\exp\left(\frac{hc}{k_B T \lambda}\right) - 1\right]^2$.

$$0 = \left(-5 + \frac{hc}{k_B T \lambda}\right) \exp\left(\frac{hc}{k_B T \lambda}\right) + 5$$

Let

$$x = \frac{hc}{k_B T \lambda}. \quad (2)$$

Then

$$0 = (-5 + x)e^x + 5.$$

Solving this equation numerically yields $x \approx 4.96511$. Equation (2) then becomes

$$4.96511 \approx \frac{hc}{k_B T \lambda}.$$

Therefore, setting $\lambda = \lambda_{\max}$ and solving for $\lambda_{\max} T$, we get the Wien displacement law.

$$\lambda_{\max} T \approx \frac{hc}{(4.96511)k_B} \approx \frac{(6.626 \times 10^{-34} \text{ J} \cdot \text{s}) \left(299\,792\,458 \frac{\text{m}}{\text{s}}\right)}{4.96511(1.38 \times 10^{-23} \text{ J} \cdot \text{K}^{-1})} \approx 2.90 \times 10^{-3} \text{ m} \cdot \text{K}$$